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- Aqueous ink composition.
- The present invention provides an aqueous ink composition which has excellent physical properties in all aspect.

An aqueous ink composition comprising

- (a) an aqueous medium and
- (b) and azo dye which has a hydroxyl group and/or an amino group, and also has 1 to 4 phosphorus-containing groups represented by the formula;

$$\frac{(O)_{n}}{(X)_{m}} \stackrel{[O]}{=} OM$$

wherein X presents an oxygen atom or an alkylene group having 1 to 4 carbon atoms, M represents a hydrogen atom, an alkyl group having 1 to 3 carbon atoms or a hydroxyalkyl group having 1 to 4 carbon atoms), m represents 0 or 1, n represents 0 or 1.

AQUEOUS INK COMPOSITION

The present invention relates to an aqueous ink composition which is excellent in writing and recording properties (especially smoothness of writing and recording), penetrating and fixing properties onto paper, discharging properties, lubricating ability and stability with time.

An aqueous ink suitable for a ball-point pen and ink-jet recording requires to have adequate stability with time, discharging properties, lubricating ability, water-resistance and light resistance, in addition to good writing and recording properties and good fixing properties of written or recorded images. Conventional aqueous ink, however, has poor stability with time and lubricating properties, which often cause plugging of a pen tip and a nozzle. In case of a ball-point pen, the rotation of the ball is obstructed and a ball holder is wored to result in bad feeling in writing. In case of ink-jet recording, the plugging inhibits discharging of ink to make the recorded letters and images indistinct.

In order to obviate the above mentioned problems, many improvements have been done, for example, an aqueous ink containing a polyalkylene glycol derivative

(JP-A- 65608/1979), an aqueous ink containing an unsaturated fatty acid

(JP-B- 3718/1982), an aqueous ink containing a phosphoric acid derivative

15 (JP-A-5773/1982 and 28472/1985) and an aqueous ink containing a polyoxyethylene fatty acid ester or a polyoxyethylene higher alcohol ether (JP-A-143602/1975).

These inks try to improve the defects by formulating therein a modifying agent, such as a lubricating agent, a wetting agent or a water-resisting agent. The modifying agent improves to certain extent, but the obtained inks still necessitate some improvements.

The present invention provides an aqueous ink composition which has excellent physical properties in all aspect.

An aqueous ink composition comprising

(a) an aqueous medium and

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(b) an azo dye which has a hydroxyl group and/or an amino group, and also has 1 to 4 phosphoruscontaining groups represented by the formula:

$$-(X)_{\overline{m}} \xrightarrow{[]{O}_{OM}} P$$

wherein X presents an oxygen atom or an alkylene group having 1 to 4 carbon atoms, M represents a hydrogen atom, an alkali metal atom or -NH(R)₃ (wherein R, which is the same or different, represents a hydrogen atom, an alkyl group having 1 to 3 carbon atoms or a hydroxyalkyl group having 1 to 4 carbon atoms), m represents 0 or 1, n represents 0 or 1.

The aqueous medium of the present invention is water or a mixture of water and one or more other water-soluble organic solvents, especially polar solvents. Typical examples of the organic solvents are monoalcohols, such as methanol, ethanol and propanol; polyhydric alcohols, such as ethylene glycol, diethylene glycol, triethylene glycol, tetraethylene glycol, glycerol, propylene glycol and dipropylene glycol; hydroxyethers, such as ethyleneglycol monomethyl ether, methyleneglycol monoethyl ether, diethyleneglycol monomethyl ether, diethyleneglycol monomethyl ether, and propyleneglycol monoethyl ether; hydroxyether esters, such as ethyleneglycol monomethyl ether acetate and diethyleneglycol monoethyl ether acetate; formamide; N,N-dimethylformamide; dimethylsulfoxide and N-methyl-2-pyrrolidone.

The azo dye of the present invention should have 1 to 4 phosphorus-containing groups represented by the formula [I]:

$$-(x)_{\overline{m}} \xrightarrow{(O)_{\overline{n}} \atop OM} [I]$$

In the formula [I], X represents an oxygen atom or an alkylene group having 1 to 4 carbon atoms, M

represents a hydrogen atom, an alkali metal atom (e.g. a lithium atom, a sodium atom or a potassium atom) or -NH(R)₃ (wherein R, the same or different, represents a hydrogen atom, an alkyl group having 1 to 3 carbon atoms or a hydroxyalkyl group having 1 to 4 carbon atoms), m represents 0 or 1, n represents 0 or 1.

Typical examples of the group [I] are

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The azo dye should also have a hydroxyl group and/or an amino group, in addition to the phosphorus-containing group [1]. Number of the hydroxyl group and the amino group is not limited, but generally 1 to 4. The azo dye may also contain -SO₃M and/or -COOM wherein M is the same as mentioned above. Typical examples of the azo dyes are acid azo dyes and direct azo dyes. Preferred are monoazo dyes, disazo dyes and trisazo dyes.

The azo dye can be prepared in a method known to the art. For example, the azo dye in which M in the formula [i] is a hydrogen atom is prepared by a diazo-coupling reaction of a diazo component having a phosphonic acid group and a coupling component. Examples of the diazo components having a phosphonic acid group are aminobenzenephosphonic acid, aminonaphthalenephosphonic acid, amino-phenylazonaphthalenephosphonic acid, amino-(phosphonophenyl)azonaphthalene and derivativesthereof. Examples of the coupling components are phenols, such as phenol, cresol and dioxybenzene; naphthols, such as naphthol and oxynaphthalenesulfonic acid; aminophenols; aminonaphthols, such as 1-amino-8-naphthol-3,6-disulfonic acid, 2-amino-5-naphthol-7-sulfonic acid, 2-amino-8-naphthol-6-sulfonic acid and 1-amino-8-hydroxy-3-phosphono-6-naphthalenesulfonic acid; naphthylamines, such as naphthylamine and aminonaphthalenesulfonic acid; pyrazoles; pyrozolones; N-(phosphonoalkyl)-anilines; and derivatives thereof.

In case where M in the group [I] is not a hydrogen atom, the azo dye thus obtained may be prepared by treating the above azo dye with an alkali metal hydroxide, an alkylamine having 1 to 3 carbon atoms or an alkanolamine having 1 to 4 carbon atoms in water or an aqueous medium at a pH value of at least 9, preferably at about pH 10. Examples of the alkali metal hydroxides are sodium hydroxide, potassium hydroxide and lithium hydroxide.

Examples of the alkylamines are methylamine, dimethylamine, trimethylamine, ethylamine, diethylamine, triethylamine, methylethylamine, n-propylamine and isopropylamine. Typical examples of the alkanolamines are ethanolamine, diethanolamine, triethanolamine and propanolamine.

Representative examples of the azo dye of the present invention are:

OH

Na0
$$P=0$$

Na0 $P=0$

Na0 $P=0$

Azo dye 2 (red)

$$\begin{array}{c|c}
& OH \\
& N=N \\
& OH \\
& N=N \\
& OH \\
& NH2
\end{array}$$

Azo dye 3 (black)

$$O = P \xrightarrow{ONa} CH_2 SO_3Na$$
OH
$$O = N = N \xrightarrow{OCH_3} OH$$

$$O = N = N \xrightarrow{OCH_3} OH$$

Azo dye 4 (black)

$$\begin{array}{c}
0 \\
P \longrightarrow N = N \longrightarrow N = N \longrightarrow NH_{2} \\
0 0 \\
Li Li
\end{array}$$

$$\begin{array}{c}
0 \\
0 \\
SO_{3}Li
\end{array}$$

$$\begin{array}{c}
0 \\
0 \\
0 \\
0
\end{array}$$

Azo dye 5 (black)

$$H_{2}N \xrightarrow{\text{CH}_{3}} H_{2} \xrightarrow{\text{CH}_{3}} H_{2} \xrightarrow{\text{OH}} 0 = P \xrightarrow{\text{ONa}} H_{2}N \xrightarrow{\text{NH}_{2}} H_{2} \xrightarrow{\text{NH}_{2}} H_{3} \xrightarrow{\text{CH}_{3}} H_{2} \xrightarrow{\text{NH}_{2}} H_{3} \xrightarrow{\text{NH}_{2$$

Azo dye 6 (black)

$$NaO = 0$$
 $NaO = 0$
 NaO

Azo dye 7 (black)

Na0
$$P = 0$$
Na0 $P = 0$
CH3 CH3 NH2 OH $O = P < ONa$
Na0 $P = 0$
Na0 $P = 0$
S03Na $O = P < ONa$
ONa

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Azo dye 8 (black)

$$\begin{array}{c}
0 \\
P - CH_2 - O - N = N - O - N =$$

Azo dye 9 (purple)

Azo dye 10 (blue)

$$Na0 > P \longrightarrow N = N \longrightarrow N = N \longrightarrow SO_3Na SO_3Na$$

Azo dye ll (blue)

Azo dye 12 (black)

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Azo dye 13 (black)

Azo dye 14 (black)

Azo dye 15 (black) and

Azo dye 16 (black)

The above azo dye can be used in combination.

An amount of the azo dye is not limited, depending upon its usage and the like. For a ball-point pen, the dye can be present in the aqueous ink composition in an amount of 3 to 15 % by weight, preferably 5 to 10 % by weight. For ink-jet, it can be present in an amount of 1 to 10 % by weight, preferably 1 to 3 % by weight.

Beside the above mentioned components, the aqueous ink composition of the present invention may contain additives, such as surfactants, surface tension controlling agents, wetting agents, viscosity modifiers, pH controlling agents, antiseptic agents and corrosion preventive agents.

The surfactants can be anionic, cationic, or nonlonic. Typical examples of the surfactants are a fluorine-containing surfactant (e.g. Unidine DS-401 and DS-402 available from Daikin Industries, Ltd., and Megafac F-1420 and F-177 available from Dainippon Ink and Chemicals, Inc.), a silicon-containing surfactant (e.g. Silicon L-7607 available from Nippon Unicar Co., Ltd.) and a phosphate surfactant.

Typical examples of the surface tension controlling agents are polyethyleneglycol ether sulfuric acid or an ester thereof, a poly(2-vinylpyridine) derivative, polyoxyethylene alkyl ether and polyoxyethylene alkyl ester.

Typical examples of the wetting agents are polyalkylene glycols, such as polyethylene glycol and polypropylene glycol; alkylene glycols, such as ethylene glycol, propylene glycol, butylene glycol and hexylene glycol; alkyleneglycol monoalkyl ethers, such as diethyleneglycol monomethyl ether, diethyleneglycol monoethyl ether, ethyleneglycol monomethyl ether and ethyleneglycol monoethyl ether; and glycerol.

Typical examples of the viscosity modifiers are polyvinylalcohol, polyvinylpyrolidone, hydroxyethylcellulose and water-soluble acryl resins.

Typical examples of the pH controlling agents are alkali metal hydroxides, such as sodium hydroxide, potassium hydroxide and lithium hydroxide; lower alkanolamines, such as ethanolamine, diethanolamine and propanolamine; and ammonium hydroxide.

Typical examples of the antiseptic agents are phenol and sodium benzoate. Typical examples of the corrosion preventive agents are benzotriazole and ethylenediaminetetraacetate.

The lnk composition of the present invention is excellent in lubricating ability and therefore good in writing and recording properties, especially smoothness of writing and recording. When it is applied to ink-jet, a discharging amount is constant, which makes it possible to write or record letters and images more clearly. Also, the composition dissolves dyes well and is stable with time, and therefore substantially no plugging occurs. Writing and recording can be done in good and stable feeling.

Further, since the dye of the present invention has a hydroxyl group and/or an amino group in addition to the phosphorus containing group, penetrating and fixing properties onto paper of the lnk composition is superior to the conventional one and enhances color value and recording rate.

It is believed that the interaction of the phosphorus-containing group, basic compounds and solvents enhances the solubility of dye. It is believed that the plugging occurs by the formation of insoluble materials which are produced by reacting the metal ions (e.g. Cu^2 , Mg^2 , Fe^2 , etc.) in the ink with carbon dioxide in the air. In the present invention, since the phosphorus-containing group of the azo dye is chemically bonded with the metal ions in preference to the reaction of carbon dioxide produces insoluble materials, the plugging is effective prevented. Further, the dye itself acts as lubricating agent and therefore improves the physical properties.

EXAMPLES

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The present invention is illustrated by the Examples which, however, are not to be construed as limiting to the present invention to their details.

Examples 1 to 15

The components shown in Table 1 were dissolved at 40 to 50 °C and filtered with a membrane filter having a pore size of 0.8 micrometer to obtain ink compositions 1 to 15 for a ball-point pen.

The lnk was charged in a ball-point pen and subjected to an evaluation test of smoothness of writing and feeling of writing by using a writing tester. The results are shown in in Table 1.

Comparative Examples 1 to 9

Ink compositions 1' to 9' were prepared as generally described in Examples 1 to 15 using the components shown in Table 1 and the same evaluation was conducted. The results are shown in Table 1.

In Table 1, the other dye compounds (a to i) are compounds having the following formula;

Dye compound a (red)

Dye compound b (red)

$$SO_3Li SO_3Li$$

$$SO_3Li$$

$$SO_3Li$$

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Dye compound c (black)

OCH3 OH

OCH3 OH

SO₃L i CH₃ SO₃Na

Dye compound d (purple)

LiO₃S-O-N=N-O-NH₂
SO₃Li

Dye compound e (black)

 $^{50}2^{\text{Na}} \xrightarrow{\text{CH}_3} \xrightarrow{\text{CH}_3} \xrightarrow{\text{NH}_2 \text{OH}} \xrightarrow{\text{SO}_3 \text{Na}}$ 40 $^{$

Dye compound f (black)

Dye compound g (blue)

Dye compound h (black)

$$\begin{array}{c} \text{NH}_2 \\ \text{H}_2 \text{N} - \bigcirc \\ \text{N} = \text{N}$$

Dye compound i (black)

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	_		_							_																	
. 5				(parts by weight)	Other dye compound										-		(parts of veight)	present invention	The azo dye of the								
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20																					10				-	-	
		L																		ā					5	55	
	L																		10						6	6	
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	•	318	113	*	Feeling writing	Written distance									
	tabi	8	2	Bu=1	g of g (3)										- 1
	•	according to JIS S 6038.	₩	9-Black, Bu-Blue,	2"	(2)									١
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	B : Smoothness is inferior	B 1 1000~ 1300g	UB 1		₩	>	8	75	_	0.2	s	10	7	7	Examples
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	both	9	2		C	8	8	75	_	0.2	5	10	12	12	
	C: Smoothness and stability		Y# 11		<u> </u>	^	8	75	1	0.2	5	10	13	13	
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		500	e fr		С	>	В	75		0.2	5	10	15	15	
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	144	ŏ	e i k		B	В	20	75		0.	5	10	2	2	
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	10		18		D	C	₩	75	-	0.2	5	10	-	-	•
	•11		ā		a	. 8		75	-	0.	5	10	6	9	

Example 16

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Two parts by weight of the compound 7 was dissolved in 75 parts by weight of deionized water, to which 10 parts by weight of diethylene glycol and 0.1 part by weight of sodium benzoate were added. The mixture was mixed at 40 to 50 °C to form a uniform solution which was then filtered with a membrane filter having a pore size of 0.8 micrometer and deaerated to obtain an ink composition for ink-jet.

The ink was charged in a vinyl chloride container and subjected to a recording test for a long period of time using an ON-Demand Type recording machine available from Seiko Epson Co., Ltd. as HG-2000. A feed rate of the ink come up with a recording rate, and the ink kept stable and no plugging of a nozzle was observed. Accordingly, recording conditions were constant and stable.

Comparative Example 10

An ink composition was prepared as generally described in Example 16, with the exception that the dye f was employed instead of the compound 7 and the same test was conducted. Discharge of the ink was not uniform to result in unclear recorded images.

Claims

- 1. An aqueous ink composition comprising
 - (a) an aqueous medium and
- (b) an azo dye which has a hydroxyl group and/or an amino group, and also has 1 to 4 phosphoruscontaining groups represented by the formula;

$$(0)_{n} OM$$

$$(X)_{m} P OM$$
[1]

wherein X presents an oxygen atom or an alkylene group having 1 to 4 carbon atoms, M represents a hydrogen atom, an alkali metal atom or -NH(R)₃ (wherein R, which is the same or different, represents a hydrogen atom, an alkyl group having 1 to 3 carbon atoms or a hydroxyalkyl group having 1 to 4 carbon atoms), m represents 0 or 1, n represents 0 or 1.

2. The ink composition according to Claim 1 wherein the azo dye is a compound of the formula

$$\begin{array}{c} \text{RO} \\ \text{HO} \\ \text{D} \\ \text{OH} \\$$

$$\begin{array}{c} \text{OH} \\ \text{NaO} \\ \text{HO} \end{array} \rightarrow P = 0 \\ \text{NaO} \rightarrow P = 0 \\ \text{SO}_{3}N_{a} \\ \text{SO}_{3}N_{a}$$

$$\begin{array}{c|c}
 & OH \\
 & O$$

$$O-N-N-N-N-N-N-N-N+2$$

$$O-PONa CH2 SO3Na$$

$$\begin{array}{c} \text{NaO} \\ \text{NaO} \\ \text{P=O} \\ \text{O} \\ \text{CH}_3 \\ \text{CH}_3 \\ \text{NH}_2 \\ \text{OH} \\ \text{OH} \\ \text{ONa} \\ \text{SO}_3 \\ \text{Na} \\ \text{SO}_3 \\ \text{Na} \\ \text{SO}_3 \\ \text{Na} \\ \text{Na} \\ \text{ONa} \\ \text{ONa}$$

NaO P = 0
NaO I CH3 CH3 NH2 OH O = P
$$\stackrel{\text{ONa}}{\text{ONa}}$$

H2N $\stackrel{\text{O}}{\text{O}}$ N = N $\stackrel{\text{O}}{\text{O}}$ O = P $\stackrel{\text{ONa}}{\text{ONa}}$

$$\begin{array}{c}
0 \\
P - CH_2 \\
OK OK
\end{array}$$

$$\begin{array}{c}
N = N - OH \\
NHCOCH_3 \\
SO_3K
\end{array}$$

$$\begin{array}{c}
SO_3K \\
SO_3K
\end{array}$$

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- 3. The ink composition according to Claim 1 or 2 wherein said azo dye is present in the ink composition in an amount of 3 to 15 % by weight for a ball-point pen.
- 4. The ink composition according to Claim 1 or 2 wherein the azo dye is present in an amount of 1 to 10 % by weight for ink-jet recording.
 - 5. An ink for a ball-point pen comprising the composition according to any one of Claims 1 to 3 and optionally usual additives.

	6. An ink optionally usu	for ink-jet rec Jai additives.	ording compri	ising the comp	osition according	to any one of	Claims 2 or 4 and
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